Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

## Mostafa M. Amini,<sup>a</sup> Amirreza Azadmeher<sup>a</sup> and Seik Weng Ng<sup>b</sup>\*

<sup>a</sup>Department of Chemistry, Shahid Beheshti University, Tehran, Iran, and <sup>b</sup>Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia

Correspondence e-mail: seikweng@um.edu.my

#### **Key indicators**

Single-crystal X-ray study T = 173 K Mean  $\sigma$ (C–C) = 0.004 Å Disorder in main residue R factor = 0.028 wR factor = 0.089 Data-to-parameter ratio = 18.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## catena-Poly[[ethyldiphenyltin(IV)]- $\mu$ -isonicotinato- $\kappa^2 O:N'$ ]

In the title compound,  $[Sn(C_2H_5)(C_6H_5)_2(C_6H_4NO_2)]_n$ , the isonicotinate group functions in an N,O-bridging mode to link adjacent triorganotin groups into an infinite chain. The metal centre shows  $SnC_3NO$  trigonal-bipyramidal coordination, with the N and O atoms in axial positions.

Received 15 August 2006 Accepted 17 August 2006

## Comment

This study continues our work on carboxylic acid derivatives of ethyldiphenyltin(IV). The previous study (Amini *et al.*, 2006) described the chloroacetate derivative, which exists as a carboxylate-bridged chain compound. The carboxylic acid used in the present study is isonicotinic acid. This acid was also used in the synthesis of the triphenyltin derivative, reported by Ng & Kumar Das (1992); its crystal structure features bridging through the carboxylate O and pyridyl N atoms, to give rise to a zig-zag chain structure. The dative Sn-N bond is, however, relatively long [2.582 (3) Å], so that the geometry about Sn is severely distorted.



In the title compound, (I), replacing the phenyl group by an ethyl group leads to no significant difference in the SnC<sub>3</sub>NO trigonal-bipyramidal coordination environment (Table 1; Fig. 1). The compound adopts a linear bridged motif, and the Sn-N bond is similarly weak [2.546 (2) Å]. The very different carboxylate C15-O1 and C15-O2 bond lengths in (I) imply charge localisation for this group.

### **Experimental**

Ethyldiphenyltin iodide was prepared by the method of Davison & Rakita (1970). This reagent (0.86 g, 2 mmol) was then treated with sodium isopropoxide (0.16 g, 2 mmol) in propan-2-ol (20 ml) to give ethyldiphenyltin isopropoxide and sodium iodide. The sodium iodide precipitate was removed by filtration. Isonicotinic acid (0.24 g, 2 mmol) was added to the filtrate and evaporation of the solvent gave a white solid, which was purified by recrystallization from a dichloromethane–methanol (1:1  $\nu/\nu$ ) mixture at 278 K to furnish colourless crystals of (I) (m.p. >573 K).

© 2006 International Union of Crystallography All rights reserved

# metal-organic papers

V = 912.97 (2) Å<sup>3</sup>

 $D_x = 1.543 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

 $\mu = 1.41 \text{ mm}^{-1}$ 

T = 173 (2) K

 $R_{\rm int} = 0.029$ 

 $\theta_{\rm max} = 27.5^{\circ}$ 

Plate, colourless

 $0.38 \times 0.20 \times 0.10 \ \mathrm{mm}$ 

13345 measured reflections

 $w = 1/[\sigma^2(F_0^2) + (0.061P)^2]$ 

where  $P = (F_0^2 + 2F_c^2)/3$ 

+ 0.7851P]

 $(\Delta/\sigma)_{\rm max} = 0.001$ 

 $\Delta \rho_{\rm max} = 2.36 \text{ e } \text{\AA}^{-3}$ 

 $\Delta \rho_{\rm min} = -1.10 \text{ e} \text{ Å}^{-3}$ 

4146 independent reflections 3983 reflections with  $I > 2\sigma(I)$ 

Z = 2

#### Crystal data

$$\begin{split} & [\mathrm{Sn}(\mathrm{C}_{2}\mathrm{H}_{5})(\mathrm{C}_{6}\mathrm{H}_{5})_{2}(\mathrm{C}_{6}\mathrm{H}_{4}\mathrm{NO}_{2})] \\ & M_{r} = 424.05 \\ & \mathrm{Triclinic}, \ P\overline{1} \\ & a = 9.7562 \ (1) \ \mathring{\mathrm{A}} \\ & b = 9.9711 \ (1) \ \mathring{\mathrm{A}} \\ & c = 10.2908 \ (1) \ \mathring{\mathrm{A}} \\ & \alpha = 78.990 \ (1)^{\circ} \\ & \beta = 77.066 \ (1)^{\circ} \\ & \gamma = 70.622 \ (1)^{\circ} \end{split}$$

#### Data collection

Bruker APEXII CCD area-detector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\rm min} = 0.616, T_{\rm max} = 0.872$ 

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.028$   $wR(F^2) = 0.089$  S = 1.124146 reflections 227 parameters H-atom parameters constrained

#### Table 1

Selected geometric parameters (Å, °).

| Sn1-C1                 | 2.155 (3) | Sn1-O1                  | 2.212 (2) |
|------------------------|-----------|-------------------------|-----------|
| Sn1-C7                 | 2.186 (3) | C15-O1                  | 1.313 (3) |
| Sn1-C13                | 2.168 (3) | C15-O2                  | 1.243 (3) |
| Sn1-N1 <sup>i</sup>    | 2.546 (2) |                         |           |
| C1-Sn1-C7              | 107.9 (1) | C7-Sn1-O1               | 95.5 (1)  |
| C1-Sn1-C13             | 120.8 (1) | C7-Sn1-N1 <sup>i</sup>  | 89.1 (1)  |
| C1-Sn1-O1              | 89.3 (1)  | C13-Sn1-O1              | 93.6 (1)  |
| C1-Sn1-N1 <sup>i</sup> | 88.2 (1)  | C13-Sn1-N1 <sup>i</sup> | 84.2 (1)  |
| C7-Sn1-C13             | 130.5 (1) | N1 <sup>i</sup> -Sn1-O1 | 175.3 (1) |
|                        |           |                         |           |

Symmetry code: (i) x - 1, y, z.

The terminal C atom of the ethyl group is disordered over two sites, C14 and C14', in a 0.54 (1):0.46 (1) population ratio. The C13–C14 and C13–C14' distances were restrained to be within 0.01 Å of each other. H atoms were placed in calculated positions, with C–H = 0.93–0.97 Å, and were refined as riding, with  $U_{iso}(H) = 1.2U_{eq}(C)$  or  $1.5U_{eq}(methyl C)$ . The largest difference peak and deepest difference hole are within 1 Å of Sn1.

Data collection: *SMART* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve



#### Figure 1

A view of part of the polymeric chain in (I) that runs along [100]. Displacement ellipsoids are drawn at the 70% probability level and H atoms are shown as spheres of arbitrary radii. The minor disorder component is not shown. [Symmetry code: (i) x - 1, y, z.]

structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *SHELXL97*.

The authors thank the University of Canterbury, New Zealand, for the diffraction measurements. We thank the Vice-President's Office for Research Affairs at Shahid Beheshti University, and the University of Malaya, for generously supporting this work.

### References

Amini, M. M., Alijani, V., Azadmeher, A., Khavasi, H. & Ng, S. W. (2006). Acta Cryst. E62, m2028–m2029.

- Barbour, L. J. (2001). J. Supramol. Chem. 1, 189-191.
- Bruker (2004). SAINT (Version 7.12a) and SMART (Version 7.12a). Bruker AXS Inc., Madison, Wisconsin, USA.
- Davison, A. & Rakita, P. E. (1970). J. Organomet. Chem. 23, 407-436.

Ng, S. W. & Kumar Das, V. G. (1992). J. Crystallogr. Spectrosc. Res. 22, 371–374.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.